# Synthesis and Reactions of Halo-, Nitro-, and Arylazo-substituted 3-Acetyltropolones. Formation of Heterocycle-fused Troponoid Compounds

Zhi-Hong Li, Zhong-Tian Jin, and Bing-Zhu Yin

Department of Chemistry, Yanbian University, Yanii, Jilin-sheng, People's Republic of China

# Kimiaki Imafuku\*

Department of Chemistry, Faculty of Science, Kumamoto University, Kurokami, Kumamoto 860, Japan Received November 5, 1986

3-Acetyltropolone (1) reacted with bromine, iodine, and nitric acid to afford respectively 3-acetyl-5,7-dibromotropolone (2), 3-acetyl-7-iodotropolone (3), and 3-acetyl-5-nitro- (4) and 3-acetyl-5,7-dinitrotropolone (5). Azo-coupling reactions of 1 gave 3-acetyl-5-arylazotropolones 7a-f. The Schmidt reactions of 2 and 3 gave respectively 5,7-dibromo- (9) and 7-iodo-2-methyl-8*H*-cyclohept[*d*]oxazol-8-one (10), while 4 gave 3-acetamido-5-nitrotropolone (11). Compounds 2 and 4 reacted with hydroxylamine to give 3-methyl-8*H*-cyclohept[*d*]isoxazol-8-ones 12 and 13. The reactions of 2, 3, and 4 with hydrazine gave 3-methyl-1,8-dihydrocycloheptapyrazol-8-ones 15, 16, and 17.

# J. Heterocyclic Chem., 24, 779 (1987).

Recently, we reported many nucleophilic reactions of 3-acetyltropolone [1]. The reactions with nucleophiles having two reaction centers are very useful for the synthesis of heterocycle-fused troponoid compounds. On the other hand, tropolone nucleus is well-known to be susceptible to many electrophilic reactions [2]. There are many electrophilic reactions of alkyl-substituted tropolones, while the reactions of tropolones having carbonyl side-chains are little known except for 3-acetonyltropolone [3] and 4-acetyl-tropolone [4].

The present paper deals with the electrophilic substitution reactions of 3-acetyltropolone and the conversions of these products to heterocycle-fused troponoid compounds. Results and Discussion.

# Electrophilic Substitutions of 3-Acetyltropolone (1).

It is well-known that tropolones are readily brominated at the 3-, 5-, and 7-positions, predominantly at the 3-position. When 3-acetyltropolone (1) was treated with an equimolar amount of bromine, dibromo-substituted compound 2 was obtained in 34% yield, the starting material 1 recovering in 20% yield. Compound 2 was shown to be 3-acetyl-5,7-dibromotropolone from its elemental analysis  $(C_9H_6Br_2O_3)$  and spectral data. The <sup>1</sup>H nmr spectrum shows only three peaks at  $\delta$  2.66 (s, 3H) for CH<sub>3</sub>, 7.90 (d, 1H, J = 2.2 Hz) for H-6, and 8.46 (d, 1H, J = 2.2 Hz) for H-4. The reaction with two molar equivalents of bromine gave 2 in 50% yield.

Although the bromination of 4-acetyltropolone gave mono-, di-, and tri-substituted products according to the reaction conditions [4], the reactions of 3-acetyltropolone (1) gave only 5,7-dibromo-substituted compound. The reaction with N-bromosuccinimide (NBS) also afforded 2 in 21% yield as a sole product.

On the other hand, the iodination of tropolone requires the presence of carbonate. Then substitution takes place at the 3- and 7-positions. The reaction of 1 with an equimolar amount of iodine in the presence of potassium carbonate afforded only mono-substituted product 3 in 30% yield. Compound 3 was shown to be 3-acetyl-7-iodotropolone from its elemental analysis  $(C_9H_7IO_3)$  and spectral data. In the 'H nmr spectrum, the three ring protons are observed at  $\delta$  6.73 (dd, 1H, J = 10.6 and 10.2 Hz) for H-5, 7.62 (d, 1H, J = 10.6 Hz) for H-6, and 8.42 (d, 1H, J = 10.2 Hz) for H-4, besides the acetyl peak at 2.68 (s, 3H). These signals support the existence of three neighboring ring protons.

3-Acetyltropolone (1) treated with fuming nitric acid to give mononitro-substituted compound 4 in 21% yield in analogy with 4-acetyltropolone [5]. The structure was determined to be 3-acetyl-5-nitrotropolone from its elemental analysis ( $C_9H_7NO_5$ ) and spectral data. The <sup>1</sup>H nmr spectrum shows three peaks at  $\delta$  7.35 (d, 1H, J = 11.5 Hz) for H-7, 8.64 (dd, 1H, J = 11.5 and 2.5 Hz) for H-6, and 8.95 (d, 1H, J = 2.5 Hz) for H-4, besides a singlet peak at  $\delta$  2.70 (s, 3H) for CH<sub>3</sub>. The reaction of the compound 4 with sodium dithionate gave 3-acetyl-5-aminotropolone (6) in 42% yield. Further, the nitration of 1 with an excess of concentrated nitric acid gave dinitro-substituted compound 5 in 15% yield, which was 3-acetyl-5,7-dinitrotropolone. However, nitrosation of 3-acetyltropolone (1) failed and the starting material 1 was recovered.

Azo-coupling reaction of tropolone takes place at the 5-position to give crystalline dyes. The reactions of 1 with a various arenediazonium salts gave exclusively 5-arylazo-substituted 3-acetyltropolones 7a-f in good yields. Their structures were confirmed by their elemental analysis and spectral data (see: Experimental part). Compound 3 also gave 3-acetyl-7-iodo-5-(4-methylphenylazo)tropolone (8) in 56% yield. In addition, the compound 7b was converted into 3-acetyl-5-aminotropolone (6) in 32% yield by the reduction with sodium dithionate.

Figure 1

Reactions of 3-Acetyltropolone Derivatives with Hydrazoic Acid, Hydroxylamine, and Hydrazine.

The Schmidt reaction of 3- and 4-acetyltropolones gave 3-acetamidotropolone [6] and 4-aminotropolone [5], respectively. The former was hydrolyzed to afford 3-aminotropolone. This reaction was applied to 5,7-dibromosubstituted 3-acetyltropolone 2 to give cyclized product, 5,7-dibromo-2-methyl-8H-cyclohept[d]oxazol-8-one (9) [7] in 69% yield. Similarly, 3-acetyl-7-iodotropolone (3) reacted with hydrazoic acid to afford 7-iodo-2-methyl-8H-cyclohept[d]oxazol-8-one (10) in 39% yield, which was confirmed by its elemental analysis and spectral data. It is noteworthy that the Schmidt reaction of 3-acetyltropolone

derivatives gave 8*H*-cyclohept[*d*]oxazol-8-ones. However, the reaction of 3-acetyl-5-nitrotropolone (4) gave 3-acetamido-5-nitrotropolone (11) in 47% yield by the Schmidt-type rearrangement, although we previously isolated 3-acetamidotropolone in the similar manner [6]. The ir spectrum of 11 shows four typical absorptions at 3450 (NH), 3250 (OH), 1690 (acetyl C = O), and 1610 cm<sup>-1</sup> (tropone C = O).

Scheme 2

Previously, we found that 3-acetyltropolone (1) reacted with hydroxylamine and hydrazine to give respectively 3-methyl-8H-cyclohept[d]isoxazol-8-one [8] and 3-methyl-1,8-dihydrocycloheptapyrazol-8-one [9]. Then a mixture of 3-acetyl-5,7-dibromotropolone (2) and hydroxylamine in methanol was refluxed for 3 hours to give 5,7-dibromo-3methyl-8H-cyclohept[d]isoxazol-8-one (12) in 29% yield. The structure was confirmed by its elemental analysis and spectral data. The ir spectrum shows a carbonyl absorption at 1620 cm<sup>-1</sup>. In the <sup>1</sup>H nmr spectrum, a singlet peak for the methyl at  $\delta$  2.63 and two doublet peaks at  $\delta$  7.70 (J = 1.7 Hz) for H-6 and 8.62 (J = 1.7 Hz) for H-4. 5-Nitro-4 and 5-phenylazo-substituted 3-acetyltropolone also reacted with hydroxylamine to give the corresponding 8H-cyclohept[d]isoxazol-8-ones 13 and 14 in 54 and 64% yields, respectively.

When a solution of 3-acetyl-5,7-dibromotropolone (2) and hydrazine hydrate in methanol was heated under

reflux for 1 hour, 5,7-dibromo-1,8-dihydrocycloheptapyrazol-8-one (15) in 64% yield, which was confirmed by its elemental analysis and spectral data. The ir spectrum shows absorptions for the NH and C=0 groups at 3170 and 1620 cm<sup>-1</sup>, respectively. Similarly, the reactions of 7-iodo- (3) and 5-nitro-3-acetyltropolone (4) gave 3-methyl-1,8-dihydrocycloheptapyrazol-8-ones 16 and 17 in 42 and 44% yields, respectively. 3-Methyl-5-(4-methylphenylazo)-1,8-dihydrocycloheptapyrazol-8-one (18) was also obtained in a similar manner.

#### **EXPERIMENTAL**

#### Measurements.

The ir spectra were taken on a Tiantsin Guangxue WFD-7G spectrophotometer. The 'H nmr spectra were recorded with a JEOL FX-100 spectrometer.

### Bromination of 3-Acetyltropolone (1).

(a) To a stirred solution of 1 (1.6 g, 10 mmoles) and sodium acetate (1.2 g, 15 mmoles) in acetic acid (30 ml) was added dropwise a solution of bromine (1.8 g, 11 mmoles) in acetic acid (2 ml) under cooling with an icewater bath. After stirring for 3 hours, the mixture was diluted with water (30 ml). The precipitate was collected and recrystallized from methanol to afford 3-acetyl-5,7-dibromotropolone (2) as yellow needles, yield 1.1 g (34%), mp 173-174°; ir (potassium bromide): ν max 3180 (OH), 1700 (C=0), 1595 cm<sup>-1</sup> (C=0); 'H nmr (deuteriochloroform): δ 2.66 (s, 3H, CH<sub>3</sub>), 7.90 (d, 1H, J = 2.2 Hz, H-6), 8.64 (d, 1H, J = 2.2 Hz, H-4).

Anal. Calcd. for C<sub>9</sub>H<sub>6</sub>Br<sub>2</sub>O<sub>3</sub>: C, 33.57; H, 1.88. Found: C, 33.80; H, 1.90. (b) A solution of bromine (5.4 g, 40 mmoles) in acetic acid (8 ml) was added to an ice-cooled solution of 1 (3.28 g, 20 mmoles) in acetic acid (10 ml) with stirring. The mixture was stirred for 2 hours and worked up, as mentioned above, to give 2, yield 3.2 g (50%).

(c) A mixture of 1 (500 mg, 3 mmoles) and NBS (534 mg, 3 mmoles) in dioxane (5 ml) was heated on a water bath for 5 minutes. After cooling, the mixture was poured into water to afford the crystals, which were collected, washed with a sodium sulfite solution, and recrystallized from methanol to give 2, yield 200 mg (21%).

#### Iodination of 3-Acetyltropolone (1).

A stirred solution of iodine (2.24 g, 8.8 mmoles) and potassium iodide (2.24 g, 13.5 mmoles) in water (6 ml) was added dropwise into an ice-cooled solution of 1 (1.3 g, 8 mmoles) and potassium carbonate (2.44 g, 17.7 mmoles) in water (6 ml). After additional stirring for 1 hour, the mixture was washed with a sodium hydrogensulfite solution to remove an excess of iodine. The precipitate was collected and dissolved in hot water. The solution was acidified with 6M hydrochloric acid to give the crystals which were recrystallized from methanol to give 3-acetyl-7-iodotropolone (3) as greenish yellow needles, yield 700 mg (30%), mp 140-141°; ir (potassium bromide):  $\nu$  ma 3120 (OH), 1700 (C = 0), 1600 cm<sup>-1</sup> (C = 0); <sup>1</sup>H nmr (deuteriochloroform):  $\delta$  2.68 (s, 3H, CH<sub>3</sub>), 6.73 (dd, 1H, J = 10.6, H.4) H.4)

Anal. Calcd. for C<sub>9</sub>H<sub>7</sub>IO<sub>3</sub>: C, 37.27; H, 2.43. Found: C, 37.06; H, 2.34. Nitration of 3-Acetyltropolone (1).

(a) A mixture of fuming nitric acid (0.8 g, 12 mmoles) and acetic acid (2 ml) was added dropwise into a ice-cooled solution of 1 (1.6 g, 10 mmoles) in acetic acid (2 ml). After stirring for 1 hour, the precipitate was collected and recrystallized from methanol to give 3-acetyl-5-nitrotropolone (4) as yellow needles, yield 430 mg (21%), mp 138-139°; ir (potassium bromide):  $\nu$  max 3200 (OH), 1700 (C=O), 1615 cm<sup>-1</sup> (C=O); <sup>1</sup>H nmr (deuteriochloroform):  $\delta$  2.70 (s, 3H, CH<sub>3</sub>), 7.35 (d, 1H, J = 11.5 Hz, H-7),

8.64 (dd, 1H, J = 11.5, 2.5 Hz, H-6), 8.95 (d, 1H, J = 2.5 Hz, H-4). Anal. Calcd. for  $C_9H_7NO_5$ : C, 51.68; H, 3.37; N, 6.70. Found: C, 51.53; H, 3.60; N, 6.69.

(b) To an ice-cooled solution of 1 (1.6 g, 10 mmoles) in acetic acid (5 ml) was added dropwise concentrated nitric acid (d = 1.4) (2.5 ml). After stirring for 2 hours, the mixture was diluted with water (25 ml). The precipitate was collected and recrystallized from methanol to give 3-acetyl-5,7-dinitrotropolone (5) as yellow needles, yield 380 mg (15%), mp 124-125°; ir (potassium bromide):  $\nu$  max 3190 (OH), 1690 (C = O), 1600 cm<sup>-1</sup> (C = O); <sup>1</sup>H nmr (deuteriochloroform):  $\delta$  2.75 (s, 3H, CH<sub>3</sub>), 8.82 (d, 1H, J = 2.5 Hz, H-6), 8.90 (d, 1H, J = 2.5 Hz, H-4).

Anal. Calcd. for  $C_0H_6N_2O_7$ : C, 42.53; H, 2.38; N, 11.02. Found: C, 47.57; H, 2.66; N, 11.23.

#### 3-Acetyl-5-aminotropolone (6).

A suspension of 3-acetyl-5-nitrotropolone (4) (420 mg, 2 mmoles) in 10% sodium hydroxide solution (10 ml) was stirred at 50° in the presence of sodium dithionate (3.5 g). After disappearance of the color of the solution, the mixture was cooled and filtered. The filtrate was neutralized with 6M hydrochloric acid to give the precipitate. The precipitate was dissolved in water and neutralized with a saturated sodium hydrogencarbonate solution to give 3-acetyl-5-aminotropolone (6) as greenish yellow needles, yield 150 mg (42%), mp 98-99°; ir (potassium bromide):  $\nu$  max 3480 (NH), 3220 (OH), 1685 (C=O), 1610 cm<sup>-1</sup> (C=O); 'H nmr (deuteriochloroform):  $\delta$  2.58 (s, 3H, CH<sub>3</sub>), 3.47 (br, 2H, NH<sub>2</sub>), 6.8-7.0 (m, 3H), 11.72 (br, 1H, OH).

Anal. Calcd. for C<sub>9</sub>H<sub>9</sub>NO<sub>3</sub>: C, 60.33; H, 5.06; N, 7.82. Found: C, 60.47; H, 5.29; N, 8.09.

# Azo-coupling Reactions of 3-Acetyltropolone (1).

To an ice-cooled solution of 1 (820 mg, 5 mmoles) in pyridine (10 ml) was added dropwise arenediazonium chloride solution, prepared from anilines (5.5 mmoles), with stirring under cooling with an ice-water bath. After additional stirring for 2 hours, the mixture was diluted with water (10 ml) to precipitate 3-acetyl-5-arylazotropolone 7a-f as red needles (from benzene).

# 3-Acetyl-5-phenylazotropolone (7a).

This compound was obtained in a yield of 385 mg (29%), mp 144-145°; ir (potassium bromide):  $\nu$  max 3400 (OH), 1700 (C=0), 1600 cm<sup>-1</sup> (C=0). Anal. Calcd. for  $C_{15}H_{12}N_2O_3$ : C, 67.15; H, 4.51; N, 10.44. Found: C, 67.42; H, 4.36; N, 10.39.

#### 3-Acetyl-5-(4-methylphenylazo)tropolone (7b).

This compound was obtained in a yield of 884 mg (63%), mp 167-168°; ir (potassium bromide):  $\nu$  max 3400 (OH), 1700 (C = 0), 1600 cm<sup>-1</sup> (C = 0); <sup>1</sup>H nmr (deuteriochloroform):  $\delta$  2.40 (s, 3H, Ar-CH<sub>3</sub>), 2.66 (s, 3H, COCH<sub>3</sub>), 7.20 (d, 2H, J = 8.0 Hz, H-3',5'), 7.43 (d, 1H, J = 11.0 Hz, H-7), 7.74 (d, 2H, J = 8.0 Hz, H-2',6'), 8.21 (dd, 1H, J = 11.0, 2.0 Hz, H-6), 8.40 (d, 1H, J = 2.0 Hz, H-4).

Anal. Calcd. for  $C_{16}H_{14}N_2O_3$ : C, 68.07; H, 5.00; N, 9.92. Found: C, 68.22; H, 4.98; N, 9.73.

#### 3-Acetyl-5-(4-methoxyphenylazo)tropolone (7c).

This compound was obtained in a yield of 1.12 g (75 %), mp 182-183°; ir (potassium bromide):  $\nu$  max 3240 (OH), 1710 (C = 0), 1610 cm<sup>-1</sup> (C = 0). Anal. Calcd. for  $C_{16}H_{14}N_2O_4$ : C, 64.42; H, 4.73; N, 9.39. Found: C, 64.19; H, 4.69; N, 9.42.

# 3-Acetyl-5-(4-chlorophenylazo)tropolone (7d).

This compound was obtained in a yield of 992 mg (65%), mp 168-169°; ir (potassium bromide):  $\nu$  max 3400 (OH), 1700 (C = 0), 1610 cm<sup>-1</sup> [C×0]. Anal. Calcd. for  $C_{15}H_{11}ClN_2O_3$ : C, 59.31; H, 3.65; N, 9.26. Found: C, 59.33; H, 3.80; N, 9.26.

# 3-Acetyl-5-(4-bromophenylazo)tropolone (7e).

This compound was obtained in a yield of 1.46 g (84%), mp 153-154°; ir (potassium bromide):  $\nu$  max 3445 (OH), 1700 (C=0), 1610 cm<sup>-1</sup> (C=0).

Anal. Calcd. for  $C_{15}H_{11}BrN_2O_3$ : C, 51.89; H, 3.19; N, 8.07. Found: C, 51.65; H, 3.27; N, 7.89.

#### 3-Acetyl-5-(4-nitrophenylazo)tropolone (7f).

This compound was obtained in a yield of 1.0 g (64%), mp 164-165°, ir (potassium bromide):  $\nu$  max 3350 (OH), 1700 (C=0), 1610 cm<sup>-1</sup> (C=0). Anal. Calcd. for  $C_{15}H_{11}N_2O_5$ : C, 57.51; H, 3.54; N, 13.42. Found: C, 57.42; H, 3.45; N, 13.47.

#### 3-Acetyl-7-iodo-5-(4-methylphenylazo)tropolone (8).

3-Acetyl-7-iodotropolone (3) (500 mg, 2 mmoles) was treated with 4-methylbenzenediazonium chloride, prepared from p-toluidine (235 mg, 2.2 mmoles), and worked up, as mentioned above, to give 3-acetyl-7-iodo-5-(4-methylphenylazo)tropolone (8) as red needles (from benzene), yield 454 mg (56%), mp 201-202°; ir (potassium bromide):  $\nu$  max 3420 (OH), 1705 (C=O), 1600 cm<sup>-1</sup> (C=O); 'H nmr (deuteriochloroform):  $\delta$  2.46 (s, 3H, Ar-CH<sub>3</sub>), 2.73 (s, 3H, COCH<sub>3</sub>), 7.37 (d, 2H, J = 8.0 Hz, H-3',5'), 7.84 (d, 2H, J = 8.0 Hz, H-2',6'), 8.40 (d, 1H, J = 2.2 Hz, H-6), 9.20 (d, 1H, J = 2.2 Hz, H-4).

Anal. Calcd. for C<sub>16</sub>H<sub>13</sub>IN<sub>2</sub>O<sub>3</sub>: C, 47.08; H, 3.21; N, 6.86. Found: C, 47.23; H, 3.15; N, 6.71.

#### Reduction of 3-Acetyl-5-(4-methylphenylazo)tropolone (7b).

A suspension of 7b (1.0 g, 3.5 mmoles) in 10% sodium hydroxide solution (25 ml) was treated with sodium dithionate (6 g) and worked up, as mentioned above, to give 6, yield 200 mg (32%).

#### 2-Methyl-8H-cyclohept[d]oxazol-8-ones.

To a suspension of 3-acetyltropolone derivative 2 or 3 (1.5 mmoles) and sodium azide (200 mg) in chloroform (6 ml) was added dropwise concentrated sulfuric acid (1 ml) with stirring at room temperature. After additional stirring for 2 hours, the chloroform was removed by decantation. The residue was diluted with water (5 ml) to precipitate the crystals which were recrystallized from acetone to give 2-methyl-8*H*-cyclohept[*d*]oxazol-8-one 9 or 10, respectively.

#### 5,7-Dibromo-2-methyl-8H-cyclohept[d]oxazol-8-one (9).

This compound was obtained as colorless needles in a yield of 220 mg (69%), mp 202° (lit [7] 198-199°); ir (potassium bromide):  $\nu$  max 1630 cm<sup>-1</sup> (C = 0); <sup>1</sup>H nmr (deuteriochloroform):  $\delta$  2.71 (s, 3H, CH<sub>3</sub>), 8.00 (d, 1H, J = 1.7 Hz, H-6), 8.57 (d, 1H, J = 1.7 Hz, H-4).

#### 7-Iodo-2-methyl-8H-cyclohept[d]oxazol-8-one (10).

This compound was obtained as colorless needles in a yield of 166 mg (39%), mp 181-182°; ir (potassium bromide):  $\nu$  max 1635 cm<sup>-1</sup> (C=O); <sup>1</sup>H nmr (deuteriochloroform):  $\delta$  2.72 (s, 3H, CH<sub>3</sub>), 6.83 (dd, 1H, J = 10.7, 9.8 Hz, H-5), 7.65 (dd, 1H, J = 10.7, 0.7 Hz, H-6), 8.69 (dd, 1H, J = 9.8, 0.7 Hz, H-4).

Anal. Calcd. for C<sub>9</sub>H<sub>6</sub>INO<sub>2</sub>: C, 36.66; H, 2.11; N, 4.88. Found: C, 36.84; H, 1.95; N, 4.78.

#### 3-Acetamido-5-nitrotropolone (11).

A suspension of 3-acetyl-5-nitrotropolone (4) (100 mg, 0.5 mmole) and sodium azide (100 mg) in chloroform (3 ml) was treated with concentrated sulfuric acid (0.5 ml) and worked up, as mentioned above, to give 3-acetamido-5-nitrotropolone (11) as yellow needles (from methanol), yield 50 mg (47%), mp 239-240°; ir (potassium bromide):  $\nu$  max 3450 (NH), 3250 (OH), 1690 (C = O), 1610 cm<sup>-1</sup> (C = O); 'H nmr (deuteriochloroform):  $\delta$  2.35 (s, 3H, CH<sub>3</sub>), 7.50 (d, 1H, J = 11.0, H-7), 8.45 (dd, 1H, J = 11.0, 1.8 Hz, H-6), 9.18 (br, 1H, NH), 10.38 (d, 1H, J = 1.8 Hz, H-4).

Anal. Calcd. for  $C_9H_8N_2O_5$ : C, 48.22; H, 3.60; N, 12.50. Found: C, 48.01; H, 3.60; N, 12.22.

#### 3-Methyl-8H-cyclohept[d]isoxazol-8-ones.

A mixture of 3-acetyltropolone derivative 2, 4, or 7a (2 mmoles) and hydroxylamine hydrochloride (279 mg, 4 mmoles) in methanol (20 ml) was refluxed for 3 hours. After cooling, the precipitate was collected and

recrystallized from ethanol to give 3-methyl-8*H*-cyclohept[*d*]isoxazol-8-one 12, 13, or 14, respectively.

#### 5,7-Dibromo-3-methyl-8H-cyclohept[d]isoxazol-8-one (12).

This compound was obtained as colorless needles in a yield of 184 mg (29%), mp 198-199°; ir (potassium bromide):  $\nu$  max 1620 cm<sup>-1</sup> (C=O); <sup>1</sup>H nmr (deuteriochloroform):  $\delta$  2.63 (s, 3H, CH<sub>3</sub>), 7.70 (d, 1H, J = 1.7 Hz, H-6), 8.62 (d, 1H, J = 1.7 Hz, H-4).

Anal. Calcd. for  $C_0H_5Br_2NO_2$ : C, 33.89; H, 1.58; N, 4.39. Found: C, 33.63; H, 1.44; N, 4.37.

#### 3-Methyl-5-nitro-8H-cyclohept[d]isoxazol-8-one (13).

This compound was obtained as yellow needles in a yield of 221 mg (54%), mp 160-161°; ir (potassium bromide):  $\nu$  max 1645 cm<sup>-1</sup> (C = O); <sup>1</sup>H nmr (deuteriochloroform):  $\delta$  2.71 (s, 3H, CH<sub>3</sub>), 7.38 (d, 1H, J = 13.4 Hz, H-7), 8.38 (dd, 1H, J = 13.4, 2.2 Hz, H-6), 8.61 (d, 1H, J = 2.2 Hz, H-4). Anal. Calcd. for  $C_9H_6N_2O_4$ : C, 52.43; H, 2.93; N, 13.39. Found: C, 52.54; H, 2.89; N, 13.28.

#### 3-Methyl-5-phenylazo-8H-cyclohept[d]isoxazol-8-one (14).

This compound was obtained as orange needles in a yield of 341 mg (64%), mp 212-213°; ir (potassium bromide):  $\nu$  max 1630 cm<sup>-1</sup> (C = 0); <sup>1</sup>H nmr (deuteriochloroform):  $\delta$  2.72 (s. 3H, CH<sub>3</sub>), 7.2-8.4 (m. 8H).

Anal. Calcd. for  $C_{15}H_{11}N_3O_2$ : C, 67.91; H, 4.18; N, 15.84. Found: C, 67.69; H, 4.02; N, 15.79.

# 3-Methyl-1,8-dihydrocycloheptapyrazol-8-ones.

A solution of 3-acetyltropolone derivative 2, 3, 4, or 7b (1.5 mmoles) and 80% hydrazine hydrate (193 mg, 3 mmoles) in methanol (10 ml) was refluxed for 1 hour. After cooling, the precipitate was collected and recrystallized from methanol to give 3-methyl-1,8-dihydrocycloheptapyrazol-8-one 15, 16, 17, or 18, respectively.

#### 5,7-Dibromo-3-methyl-1,8-dihydrocycloheptapyrazol-8-one (15).

This compound was obtained as yellow needles in a yield of 232 mg (49%), mp 274° dec; ir (potassium bromide):  $\nu$  max 3170 (NH), 1620 cm<sup>-1</sup> (C = 0); 'H nmr (deuteriochloroform):  $\delta$  2.58 (s, 3H, CH<sub>3</sub>), 7.95 (d, 1H, J = 1.5 Hz, H-6), 8.61 (d, 1H, J = 1.5 Hz, H-4), 11.21 (br, 1H, NH).

Anal. Calcd. for  $C_0H_6Br_2N_2O$ : C, 33.99; H, 1.90; N, 8.81. Found: C, 33.83; H, 1.67; N, 8.57.

# 7-Iodo-3-methyl-1,8-dihydrocycloheptapyrazol-8-one (16).

This compound was obtained as yellow needles in a yield of 182 mg (42%), mp 223-224°; ir (potassium bromide):  $\nu$  max 3220 (NH), 1610 cm<sup>-1</sup> (C=0); 'H nmr (deuteriochloroform):  $\delta$  2.61 (s, 3H, CH<sub>3</sub>), 6.54 (dd, 1H, J = 10.0, 9.8 Hz, H-5), 7.58 (d, 1H, J = 10.0 Hz, H-6), 8.63 (d, 1H, J = 9.8 Hz, H-4).

Anal. Calcd. for C<sub>9</sub>H<sub>7</sub>IN<sub>2</sub>O: C, 37.78; H, 2.47; N, 9.79. Found: C, 37.66; H, 2.35; N, 9.76.

#### 3-Methyl-5-nitro-1,8-dihydrocycloheptapyrazol-8-one (17).

This compound was obtained as yellow needles in a yield of 135 mg (44%), mp 242°; ir (potassium bromide):  $\nu$  max 3220 (NH), 1645 cm<sup>-1</sup> (C=0); <sup>1</sup>H nmr (deuteriodimethyl sulfoxide):  $\delta$  2.69 (s, 3H, CH<sub>3</sub>), 7.10 (d, 1H, J = 13.4 Hz, H-7), 8.38 (dd, 1H, J = 13.4, 2.4 Hz, H-6), 8.92 (d, 1H, J = 2.4 Hz, H-4), 14.6 (br, 1H, NH).

Anal. Calcd. for  $C_9H_7N_3O_3$ : C, 52.68; H, 3.44; N, 20.48. Found: C, 52.42; H, 3.67; N, 20.19.

# 3-Methyl-5-(4-methylphenylazo)-1,8-dihydrocycloheptapyrazol-8-one (18).

This compound was obtained as orange needles in a yield of 270 mg (65%), mp 259°; ir (potassium bromide):  $\nu$  max 3170 (NH), 1625 cm<sup>-1</sup> (C=0); 'H nmr (deuteriodimethyl sulfoxide):  $\delta$  2.42 (s, 3H, Ar-CH<sub>3</sub>), 2.63 (s, 3H, 3-CH<sub>3</sub>), 7.11 (d, 1H, J = 13.0 Hz, H-7), 7.41 (d, 2H, J = 8.4 Hz, H-3',5'), 7.83 (d, 2H, J = 8.4 Hz, H-2',6'), 8.24 (dd, 1H, J = 13.0, 1.1 Hz, H-6), 8.42 (d, 1H, J = 1.1 Hz, H-4).

Anal. Calcd. for C<sub>16</sub>H<sub>16</sub>N<sub>4</sub>O: C, 69.05; H, 5.07; N, 20.13. Found: C, 69.12; H, 4.97; N, 20.21.

## REFERENCES AND NOTES

- [1] K. Imafuku and Z.-T. Jin, Yanbian Daxue Xuebao, 35 (1983).
- [2a] T. Nozoe, "Non-benzenoid Aromatic Compounds", D. Ginsburg, ed, Interscience Publishers, New York-London 1959, pp 339-463; [b] T. Nozoe, K. Takase, and H. Matsumura, "Dai Yuki Kagaku", Vol 13, M. Kotake, ed, Asakura Shoten, Tokyo 1960, pp 1-437; [c] D. Lloyd, "Non-benzenoid Conjugated Carbocyclic Compounds", Elsevier, Amsterdam, 1984, pp 67-147.
  - [3] K. Takase, Bull. Chem. Soc. Japan, 38, 301 (1965).

- [4] K. Takase, K. Sasaki, K. Shimizu, and T. Nozoe, Bull. Chem. Soc. Japan, 44, 2460 (1971).
  - [5] K. Doi, Bull. Chem. Soc. Japan, 34, 501 (1961).
- [6] K. Imafuku, M. Furuya, and Z.-T. Jin, Bull. Chem. Soc. Japan, 57, 609 (1984).
  - [7] Y. Kitahara, Sci. Rep. Toholu Univ., I, 40, 83 (1956).
- [8] Y. Sodoh, Z.-T. Jin, K. Imafuku, and H. Matsumura, J. Heterocyclic Chem., 19, 525 (1982).
- [9] A. Yamane, M. Nagayoshi, K. Imafuku, and H. Matsumura, Bull. Chem. Soc. Japan, 52, 1972 (1979).